Syllabus for Organic Chemistry I Chemistry 221, Section 001 4.0 Semester Hours Fall Semester, 2009

Lectures: M, W, F --- 2:45 - 3:35 p.m., FH-133

<u>Discussion Section:</u> T --- 1:00 - 2:15 p.m., FH-7

Instructor: Dr. Babler (Office: FH-209-210)

e-mail: jbabler@luc.edu

Office Hours: M --- 2:00 - 2:30, 3:45 - 5:00 p.m.

W --- 3:45 – 5:00 p.m. Th --- 1:30 – 2:30 p.m. F --- 2:00 – 2:30 p.m.

Other times by appointment, please.

Prerequisites: Chemistry 106 or its equivalent (Chemistry 102 and 112).

<u>Content:</u> This course is the first semester of a comprehensive and somewhat rigorous survey of organic chemistry. The topics include: structure and bonding in organic molecules; alkanes; alicyclic compounds; stereochemistry and conformational analysis; alkyl halides; alcohols and ethers; an introduction to infrared and nuclear magnetic resonance spectroscopy and mass spectrometry in organic chemistry; alkenes; and alkynes.

<u>Learning Outcomes:</u> Students will develop an understanding of the chemical behavior of relatively simple organic molecules and the mechanisms by which such reactions occur. Students will also be introduced to the use of spectroscopy as a powerful tool for structure determination of organic molecules.

<u>Assignments:</u> Students are strongly advised to read the assigned pages in Vollhardt=s and Schore=s textbook and to work <u>some</u> of the suggested problems (listed on the attached sheet). However, there will be no formal homework assignments!

Attendance at the scheduled discussion section is strongly recommended, but not required. The instructor will generally use the time allotted for discussion section to answer students= questions and to work some of the recommended problems from the textbook.

<u>NOTE</u>: Even though attendance at the discussion section is optional, those weekly sessions are a valuable part of the course since they offer the students an opportunity to ask questions and/or request that certain topics be clarified or explained in greater detail. Due to the vast number of topics that an organic instructor is required to cover during the formal course lectures, <u>little (if any) time</u> is available to answer students= questions during the lecture section. If you do have an urgent question that arises during the lecture and you prefer not to wait until the next discussion

section, feel free to ask such questions at the end of the lecture section.

Unless you are excused by the instructor (i.e., you are repeating the course and have previously completed the lab work), students are required to perform the lab experiments and take the exams that are required for students registered for CHEM 225 (Organic Chemistry Lab A). You will do these experiments on Tuesday afternoons (2:30 -5:15 p.m.). You will not be required to attend lab on Thursday afternoons during the first several weeks of the semester --- i.e., until the formal course lectures have covered introductory topics such as structure and bonding in organic molecules, as well as nomenclature of rather simple organic structures. After that time, the Thursday afternoon (2:30 -5:15 p.m.) lab session will be used for various additional assignments. On Thursday, Sept. 10^{th} (2:30 – 5:00 p.m), the IC classroom (IC-120) has been reserved so that you can be introduced to "chemical literature searching online." You will receive training on the use of SciFinder Scholar [which provides desktop access to several databases from Chemical Abstracts Service (CAS)] and will also cover basics of "Web of Science." Additional Thursday afternoon assignments include the following: molecular modeling: conformational analysis of cyclohexane derivatives; isolation of a "terpene" from orange peels by steam distillation; characterization of the "terpene" isolated from the previous experiment by recording its IR spectrum, as well as using a polarimeter to measure its specific rotation; use of column chromatography to purify a non-polar organic compound (e.g., veratrole or anisole) contaminated with a polar compound (e.g., ethylene glycol); dehydration of an alcohol: preparation of cyclohexene; and hypochlorite oxidation of a secondary alcohol. As soon as the Tuesday afternoon laboratory work involves Apreparative experiments,@ the Thursday lab session will be used to characterize (e.g., IR, NMR, TLC) organic compounds you have prepared during experiments conducted on Tuesday afternoons.

<u>NOTE</u>: You do <u>not</u> have to register for CHEM 225; by registering for CHEM 221, you were automatically enrolled in the Tuesday afternoon lab section. Your grade for the lab work will be incorporated into your final grade for CHEM 221 (i.e., you will <u>not</u> receive a separate grade for lab work on your report card). Failure to complete this lab work will result in a grade of AI@ for CHEM 221. This grade will subsequently be changed to AF@ on your transcript unless arrangements can be made for you to complete the experimental work within six weeks of the end of the semester. Permission for students to conduct Amake-up@ experiments will be granted <u>only</u> for very serious reasons -- e.g., if a student has been hospitalized.

Grading: Three hour exams will be given during the semester, as well as a final examination.

Distribution of points:

Hour Exam I:	125
Hour Exam II:	140
Hour Exam III:	175
Lab Experiments and Exams:	210
Lecture Final Exam:	250

Total: 900 points

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Grading Scale: Exams (excluding lab exams):
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A = 550-690; B+ = 510-549; B = 440-509; B- = 400-439; C+ = 340-399; C = 280-339; C- = 230-279; D+ = 185-229; D = 140-184; F = 0-139 points.
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Laboratory work: A separate grading scale is used for lab work and will be announced by the lab coordinator for CHEM 225. Your lab grade will be adjusted prior to being incorporated into your final grade for CHEM 221. For example, if your lab grade is 90%, you will receive 189 points (90% x 210) as part of your final grade in CHEM 221.

NOTE: There is no penalty if a student misses an hour exam; instead his/her final exam score will be used to determine a larger % of the final grade. For example, if one misses the first hour exam, his/her final examination score would represent 41.7% (i.e., 375 points) of the final grade. No Amake-up@ hour exams will be administered, unless your absence is due to your required participation in a University-sanctioned event.

WARNING: Since you are allowed to use an 8.0 x 10.5" sheet of notes as an aid during the final exam for the lecture section of this course, some of you may be tempted to skip the other tests. Despite the availability of such notes during the final exam, it usually proves to be quite difficult; and therefore you should miss an exam only in case of <u>serious</u> illness and the like!

Hour exam I will be given on Friday, September 18; hour exam II will be administered on Friday, October 16; and the third hour exam is scheduled for Friday, November 20. The final exam for this course (lecture section only) is to be given on Friday, December 11, 1:00 p.m. – 3:00 p.m. in FH-133.

NOTE: The Chemistry Department administers make-up final exams (<u>different</u> from the regular final exam) to those students who have a legitimate excuse (e.g., death in the immediate family; <u>serious</u> illness -- which does not include organophobia!; a court appointment that cannot be rescheduled, etc.). If for one of the latter reasons you are unable to take the regularly scheduled final exam, please inform the instructor (e-mail: <u>jbabler@luc.edu</u>) <u>promptly</u> --- but <u>no later</u> than 48 hours after the date of the final exam --- so that a final grade of "I" can be assigned to you.

NOTE: A grade of "I" will <u>not</u> be assigned to you unless you can verify that there was a valid reason for your missing the final exam (e.g., hospitalization or death in the immediate family). Even if your final grade is an "I," you are still required to take a "make-up" final exam or the "I" will be converted to an "F" by the Dean's office.

NOTE: Oversleeping, forgetting what day it is, etc. are not valid excuses for missing the final exam. The Chemistry Dept. will <u>not</u> accept such explanations from students, and a grade of A0@ will be assigned for your final exam score!

The last day to withdraw from class with a grade of AW@ is Friday, October 30. After that date, the Dean=s office will automatically assign the grade of AWF@ when a student withdraws from the course (except for cases in which the student is hospitalized or encounters some very <u>serious</u>

difficulty.) **NOTE**: In regard to advice concerning a decision to withdraw (or not) from CHEM 221, each student will meet briefly with the instructor sometime during the week of October 19-23 to discuss his/her progress in the course. If you have taken neither of the two tests given in the course at this stage of the semester, you are <u>strongly</u> advised to withdraw from the class.

Textbooks:

Laboratory: In order to conduct the Tuesday afternoon lab experiments, you should purchase the same paperback used by the students registered for CHEM 225 --- "Catalyst Lab Manual."

To assist you in characterizing the organic compounds you prepare in lab, you may also want to purchase a copy of the following paperback: AIntroduction to Organic Spectroscopy@ (Oxford Chemistry Primers, No. 43), by L.M. Harwood, T.D.W. Claridge, and D.W. Claridge (Oxford University Press, 1997).

Lecture: Organic Chemistry: Structure and Function, Fifth Edition, by Vollhardt and Schore. Recommended: A copy of the study guide and solutions manual for Vollhardt=s and Schore=s textbook (available at the bookstore).

Required: <u>Darling Models</u>, Kit #1A (See: <u>www.molecularvisions.com</u>) or a similar molecular models kit (available at the bookstore).

<u>NOTE</u>: As a possible study aid, you may want to consider use of the following: A **QuickFacts Study Card** is included with every copy of the Vollhardt/Schore text --- located at the back of the book on green paper. With page references included for easier studying, this helpful resource should be more useful than "flash cards" or unrelated study tools.

Another possible study aid is a paperback by D. R. Klein entitled "Organic Chemistry as a Second Language: Translating the Basic Concepts" (published in 2004 by John Wiley & Sons, Inc.; ISBN 0-471-27235-3; www.wiley.com/college/klein). The goals of the latter book are to help the student to develop the skills required to solve a variety of problem types in organic chemistry and to point out the fundamental principles in organic chemistry.

An additional study aid is a paperback by D.P. Weeks entitled "Pushing Electrons: A Guide for Students of Organic Chemistry," Third Edition (Thomson Brooks/Cole); ISBN 0-03-020693-6. The first 3 chapters (pp. 1-161) of this workbook are intended to help a student understand "structure and bonding in organic molecules," as well as techniques of "electron pushing" so as to comprehend reaction mechanisms.

Supplementary Textbooks:

Organic Chemistry, Ninth Edition, by T.W.G. Solomons and C. Fryhle (John Wiley & Sons, 2008).

Organic Chemistry, by J. McMurry, Seventh Edition (Brooks/Cole Publishing Co., 2008).

Organic Chemistry, by F. A. Carey, Seventh Edition (McGraw-Hill, Inc., 2008). Organic Chemistry, by L.G. Wade, Jr., Seventh Edition (Pearson Prentice Hall, 2010).