

Syllabus  
Spring 2010 Chemistry 302  
Physical Chemistry II

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1. **Lecture**

TTh 10:00 a.m.–11:15 a.m.  
Damen Hall Room 641

2. **Discussion**

T 11:30 a.m.–12:45 p.m.  
Flanner Hall Auditorium

3. **Office Hours**

W 11:30 a.m.–12:45 p.m. or by appointment  
Flanner 200B

4. **Textbook and Materials**

- *Physical Chemistry: A Molecular Approach* McQuarrie and Simon (1997)
- *Problems and Solutions to Accompany Physical Chemistry: A Molecular Approach* Cox (1997)
- *Schaum's Outline of Mathematica, Second Edition* Don (2009)
- A laptop with *Mathematica* version 7, QuickTime 7, and Adobe Reader

## 5. Grading

- a) Best 2 of 3 Midterm exams (50%)
- b) Homework and Projects (25%)
- c) Cumulative Final Exam (25%)

This course is graded on a curve set to at least a 2.7 average GPA. Statistics will be presented after each exam, and so it is possible to assess *approximately* your grade. *It is impossible to know your actual score until the end of the semester.*

There are no make-up exams. If you must miss an exam, it will be entered as a 0. This will probably be the exam that you drop.

- 6. **Academic Dishonesty** The penalty for academic dishonesty is a grade of “F” for the entire course.
- 7. **Course Prerequisites** Lack of appropriate course prerequisites constitutes grounds for being administratively withdrawn from the course at any time.
- 8. **Fine Print** This course is under the auspices of the College of Arts and Sciences and the Department of Chemistry. Information that is not provided on this syllabus (e.g., registration, drop dates, final examination schedule etc.) can be found in other materials such as the Course Catalog.